

## Saponin Colorimetric Microplate Assay Kit

Cat #: orb759224 (manual)

*For Research Use Only. Not For Use in Diagnostic Procedures!*

Detection and Quantification of Saponin Content in Tissue extracts and Other biological fluids Samples.

### INTRODUCTION

Saponins are a diverse group of naturally occurring plant compounds that form stable foams when mixed with water. They include cardiac-active compounds found in digitalis and squill, as well as non-cardiac-active compounds. Cardiac-active saponins have historically been used as arrow and spear poisons by indigenous peoples in Africa and South America. Digitalis, derived from purple foxglove (*Digitalis purpurea*), was introduced into cardiac therapy in 1785 by Scottish physician William Withering. Non-cardiac-active saponins include digitonin, first identified in digitalis preparations in 1875, and dioscin, a precursor of diosgenin obtained from Mexican yam. The Saponin Colorimetric Microplate Assay Kit provides a convenient and sensitive method for detecting saponins in a variety of samples. Saponin levels are measured using a coupled chemical reaction system with a colorimetric readout at 540 nm.

### KIT COMPONENTS

Component	Volume	Storage
96-Well Microplate	1 plate	
Assay Buffer	30 ml x 4	4 °C
Dye Reagent	Powder x 1	4 °C, keep in dark
Dye Reagent Diluent	5 ml x 1	4 °C
Standard	Powder x 1	4 °C
Plate Adhesive Strips	3 Strips	
Technical Manual	1 Manual	

### MATERIALS REQUIRED BUT NOT PROVIDED

1. Sulfuric acid (98%)- Analytical Reagent Grade
2. Microplate reader to read absorbance at 540 nm
3. Pipettor, multi-channel pipettor
4. Pipette tips
5. Mortar
6. Ice-water bath

7. Timer

**REAGENT PREPARATION**

**Standard:** Briefly centrifuge prior to opening. Dissolve in 1 ml Assay Buffer to generate stock standard solution; then add 0.1 ml stock solution into 0.1 ml Assay Buffer to generate 5 mmol/L of top standard solution. Perform 2-fold serial dilutions of the top standard solution using Assay Buffer to make the standard curve. The concentration of standard curve could be 5/2.5/1.25/0.63/0.31/ 0.16/0.08/0.04 mmol/L.

**Dye Reagent:** Add 5 ml Dye Reagent Diluent to dissolve before use. Store at 4 °C in dark for 1 week after reconstitution.

**SAMPLE PREPARATION**

1. For tissue samples

Weigh out 0.1 g tissue, homogenize with 1 ml Assay Buffer, then transfer it to the microcentrifuge tubes; incubate at 50 °C water bath for 1 hour; centrifuged at 10,000g for 10 minutes, take the supernatant into a new centrifuge tube for detection.

2. For liquid samples

Detect directly.

**ASSAY PROCEDURE**

Place the 96-Well Microplate in an ice-water bath. Add following reagents into the microplate:

Reagent*	Sample**	Standard	Blank
Sample	50 µl	--	--
Standard	--	50 µl	--
Assay Buffer	--	--	50 µl
Dye Reagent	50 µl	50 µl	50 µl
Add 100 µl sulfuric acid down the inner wall slowly, mix thoroughly while keeping in the ice bath.			
Incubate at 60 °C for 15 minutes. Record absorbance measured at 540 nm after cooling.			

Note:

\*Reagents must be added sequentially and should not be premixed prior to addition.

\*\* The concentrations can vary over a wide range depending on the different samples. For unknown samples, we recommend doing a pilot experiment & testing several doses to ensure the readings are within the standard curve range.

## CALCULATION

1. Calculate the sample concentration in ASSAY PROCEDURE according to the slope of the standard curve

$$C = \frac{(\text{OD}_{\text{Sample}} - \text{OD}_{\text{Blank}}) - \text{Intercept}}{\text{Slope}} \times n \text{ (mmol/L)}$$

Calculate the initial concentration according to sample preparation procedure.

2. According to one point of the standard OD and concentration

- 2.1 According to the weight of sample

$$C = \frac{(C_{\text{Standard}} \times V_{\text{Standard}}) \times (\text{OD}_{\text{Sample}} - \text{OD}_{\text{Blank}})}{(\text{OD}_{\text{Standard}} - \text{OD}_{\text{Blank}}) \times W \times (V_{\text{Sample}} / V_{\text{Assay}})} \text{ (}\mu\text{mol/g)}$$

- 2.2 According to the volume of sample

$$C = \frac{(C_{\text{Standard}} \times V_{\text{Standard}}) \times (\text{OD}_{\text{Sample}} - \text{OD}_{\text{Blank}})}{(\text{OD}_{\text{Standard}} - \text{OD}_{\text{Blank}}) \times V_{\text{Sample}}} \text{ (}\mu\text{mol/mL)}$$

Slope: the absorbance slope of standard curve

n: the dilution factor

C<sub>Standard</sub>: the standard concentration, mmol/L =  $\mu\text{mol/mL}$

V<sub>Standard</sub>: the volume of standard in assay procedure,  $\mu\text{l}$

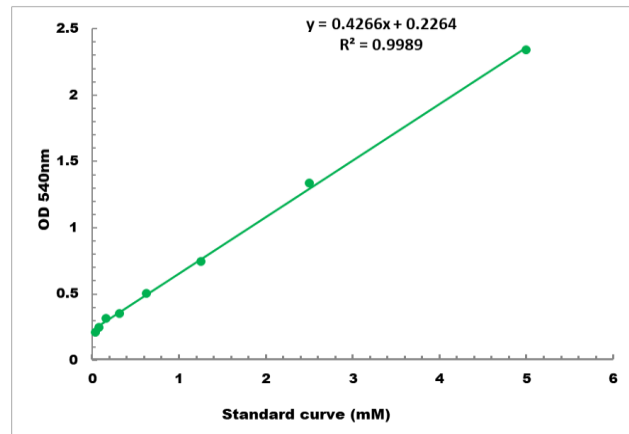
V<sub>Sample</sub>: the volume of sample in assay procedure,  $\mu\text{l}$

V<sub>Assay</sub>: the volume of Assay Buffer,  $\mu\text{l}$

W: the weight of sample, g

## TYPICAL DATA

The standard curve is for demonstration only. A standard curve must be run with each assay.



Detection Range: 0.04 mmol/L - 5 mmol/L